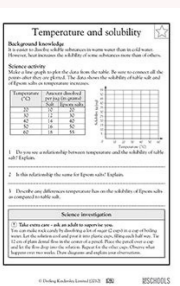
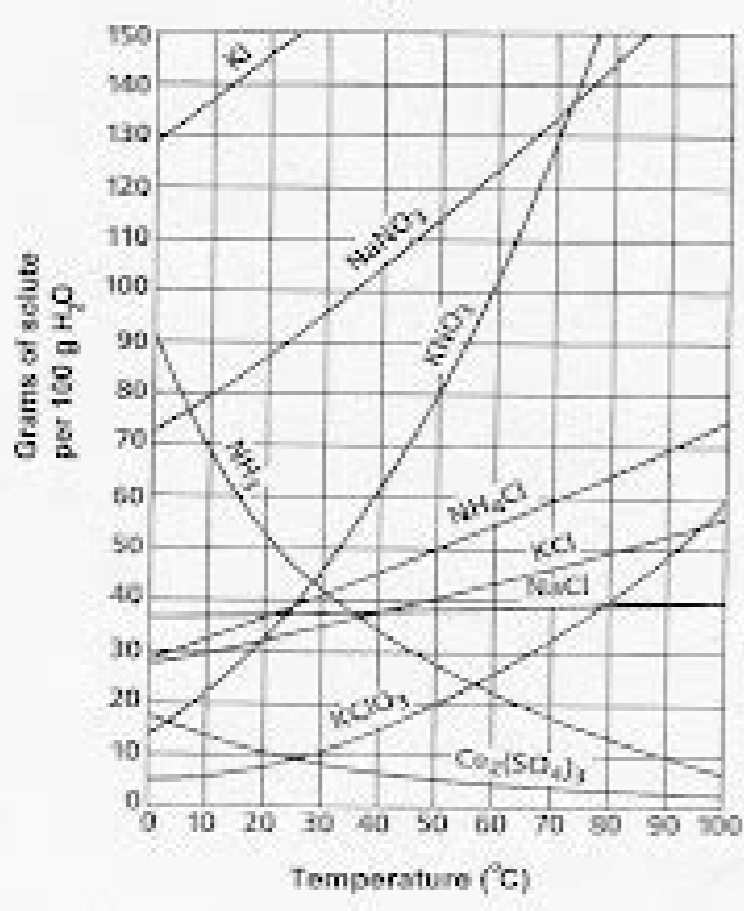


I'm not robot!

Answer the following questions based on the solubility curve below.

1. Which salt is least soluble in water at 20 °C? KClO₃
2. How many grams of potassium chloride can be dissolved in 200 g of water at 80 °C? 100g KCl (50g KCl in 100g H₂O → 100g in 200g H₂O)
3. At 40 °C, how much potassium nitrate can be dissolved in 300 g of water? 162 x 3 = 187g KNO₃
4. Which salt shows the least change in solubility from 0 °C to 100 °C? NaCl
5. At 30 °C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated, or supersaturated? unsaturated
6. A saturated solution of potassium chloride is formed from 100 g of water. If the saturated solution is cooled from 80 °C to 50 °C, how many g of precipitate are formed? 35 - 15 = 20g KCl
7. What compound shows a decrease in solubility from 0 °C to 100 °C? NH₃, Ca₂(SO₄)₃
8. Which salt is most soluble at 10 °C? KI
9. Which salt is least soluble at 50 °C? KClO₃
10. Which salt is least soluble at 90 °C? KClO₃

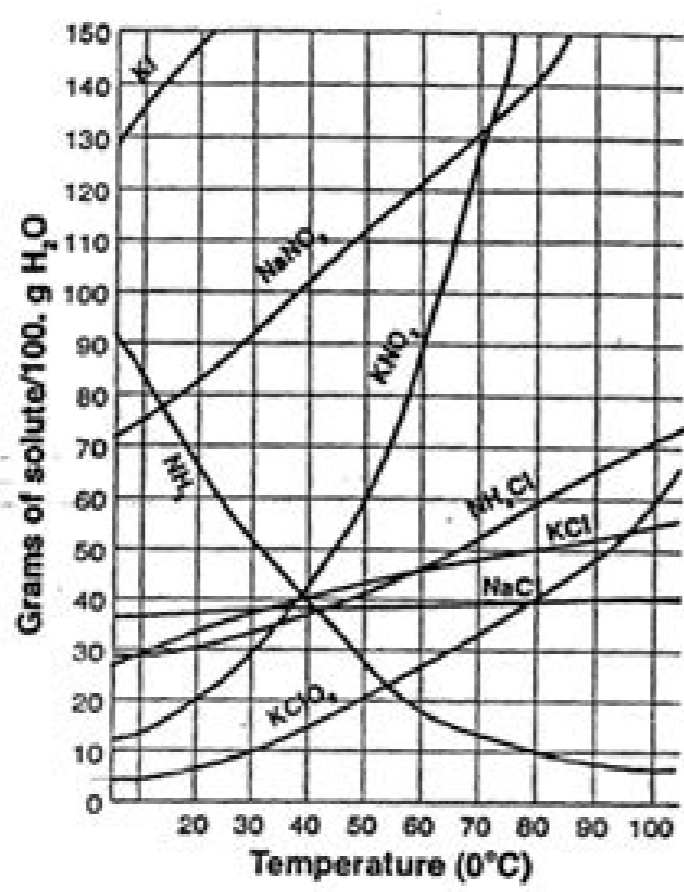


SOLUBILITY CURVES

Name: _____

Answer the following questions based on the solubility curve below.

1. Which salt is least soluble in water at 20° C? KClO₃
2. How many grams of potassium chloride can be dissolved in 200 g of water at 80° C? 100 g
3. At 40° C, how much potassium nitrate can be dissolved in 300 g of water? 123 g
4. Which salt shows the least change in solubility from 0° - 100° C? NaCl
5. At 30° C, 90 g of sodium nitrate is dissolved in 100 g of water. Is this solution saturated, unsaturated or supersaturated? unsaturated
6. A saturated solution of potassium chloride is formed from one hundred grams of water. If the saturated solution is cooled from 80° C to 50° C, how many grams of precipitate are formed? 46 - 26 g = 20 g
7. What compound shows a decrease in solubility from 0° to 100° C? NH₃
8. Which salt is most soluble at 10° C? KI
9. Which salt is least soluble at 50° C? KClO₃
- Which salt is least soluble at 90° C? NH₃



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